

A Facile Synthesis of 4-Aryl-2H-1-benzopyran-2-ones

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Reaction of 2-hydroxybenzophenones (**1**, **3**, **7**, **8**) with ethoxycarbonylmethylenetriphenylphosphorane affords 4-aryl-2H-1-benzopyran-2-ones (**2**, **4–6**) in excellent yields.

(Keywords: 4-Aryl-2H-1-benzopyran-2-ones; Ethoxycarbonylmethylenetriphenylphosphorane; Friedel-Crafts reaction; 2-Hydroxybenzophenones; Wittig reaction)

Eine einfache Synthese für 4-Aryl-2H-1-benzopyran-2-one

Die Reaktion der 2-Hydroxybenzophenone **1**, **3**, **7** und **8** mit Ethoxycarbonylmethylenetriphenylphosphoran ergab die 4-Aryl-2H-1-benzopyran-2-one **2** und **4–6** in ausgezeichneten Ausbeuten.

Introduction

Several 4-aryl-2H-1-benzopyran-2-ones have been reported¹ in the literature from various natural sources. Recently, some 4-aryl-2H-1-benzopyran-2-ones have been isolated from *Courtarea hexandra*² and their structure established by spectroscopic methods only. Earlier, *Perkin*^{3,4} and *Pechmann*^{5,6} condensations have been used in the synthesis of 4-aryl-2H-1-benzopyran-2-ones. In general, the yields reported by these methods are not satisfactory. In this communication, we report a facile and general method for the synthesis of 4-aryl-2H-1-benzopyran-2-ones. The method involves *Wittig* reaction^{7,8} of 2-hydroxybenzophenones with ethoxycarbonylmethylenetriphenylphosphorane [$(C_6H_5)_3P = CHCOOC_2H_5$].

Results and Discussion

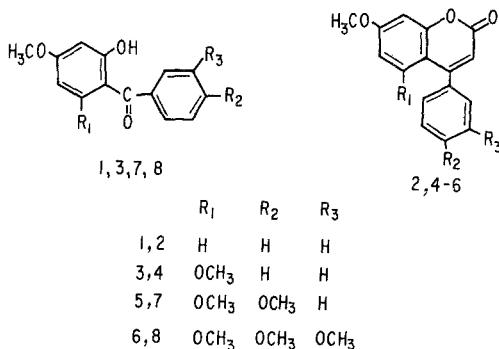
Initially, the reaction of 2-hydroxy-4-methoxybenzophenone⁹ (**1**) with ethoxycarbonylmethylenetriphenylphosphorane in anhydrous benzene

gave only one product in excellent yield (70%). The product was assigned the structure of 7-methoxy-4-phenyl-2*H*-1-benzopyran-2-one (**2**) on the basis of its IR and ¹H-NMR spectral data. The structure **2** was further confirmed by comparison with literature¹⁰ (m.p.). Similar condensation of 2-hydroxy-4,6-dimethoxybenzophenone¹¹ (**3**) with ethoxycarbonylmethylenetriphenylphosphorane in dry benzene afforded 5,7-dimethoxy-4-phenyl-2*H*-1-benzopyran-2-one (**4**). Using the same procedure, synthesis of 5,7,4'-trimethoxy-4-phenyl-2*H*-1-benzopyran-2-one (**5**) and 5,7,3',4'-tetramethoxy-4-phenyl-2*H*-1-benzopyran-2-one (**6**) has been carried out starting from 2-hydroxy-4,6,4'-trimethoxybenzophenone (**7**) and 2-hydroxy-4,6,3',4'-tetramethoxybenzophenone (**8**), respectively. The latter ketones **7** and **8** were obtained by Friedel-Craft's reaction of 1,3,5-trimethoxybenzene with 4-methoxybenzoyl chloride and 3,4-dimethoxybenzoyl chloride, respectively, in dry ether. 5,7,4'-Trimethoxy-4-phenyl-2*H*-1-benzopyran-2-one has recently been isolated² from *Courtarea hexandra*. The compounds **4** and **6** are methyl ethers of naturally occurring 5,7-dihydroxy-4-phenyl-2*H*-1-benzopyran-2-one¹² and 3',4'-dohydroxy-5,7-dimethoxy-4-phenyl-2*H*-1-benzopyran-2-one², respectively.

Table 1. Data for the

No.	Reaction time (h)	Yield (%)	M.p. (°C)	IR (KBr) γ (cm ⁻¹)
2	34	70	146–147 (116–118 ¹⁰) ^a	1 710 1 620
4	36	65	167–168 (167 ¹²) ^a	1 710 1 605
5	30	75	151–152 (151–152 ²) ^a	1 710 1 610
6	34	70	169–170 (169–170 ²) ^a	1 705 1 610
7	—	70	150–151	—
8	—	50	135–136	—

^a Lit. m.p.



All these coumarins exhibited a C=O band in their IR spectra between 1710 and 1705 cm⁻¹. The ¹H-NMR spectral data were in agreement with the proposed structures. The yields and spectroscopic data are summarised in Table 1.

Neutral conditions, high yield and ready availability of the phosphorane enhances the utility of the method.

compounds 2 and 4-8

¹H-NMR (CDCl₃) TMS,
 δ (ppm)

3.77 (s, 3 H, OMe), 5.90 (s, 1 H, H-3), 6.62–6.81
(m, 2 H, H-6 and H-8), 7.07–7.31 (m, 5 H, C₆H₅),
7.43 (d, J = 9 Hz, 1 H, H-5)
3.70 (s, 3 H, OMe), 3.83 (s, 3 H, OMe), 5.84 (s, 1 H,
H-3), 6.27 (d, J = 2.5 Hz, 1 H, H-6), 6.47 (d, J = 2.5 Hz,
1 H, H-8), 7.13–7.40 (m, 5 H, C₆H₅)
3.50 (s, 3 H, OMe-5), 3.89 (s, 6 H, OMe-7, OMe-4'),
5.85 (s, 1 H, H-3), 6.14 (d, J = 2.5 Hz, 1 H, H-6),
6.45 (d, J = 2.5 Hz, 1 H, H-8), 6.84 (d, J = 9 Hz, 2 H,
H-3', H-5'), 7.17 (d, J = 9 Hz, 2 H, H-2', H-6')
3.47 (s, 3 H, OMe-5), 3.86 and 3.92 (each s, 6 H and
3 H, OMe-7, OMe-3', OMe-4'), 6.01 (s, 1 H, H-3),
6.25 (d, J = 2.5 Hz, 1 Hz, H-6), 6.52 (d, J = 2.5 Hz, 1 H,
H-8), 6.79–6.94 (m, 3 H, H-2', H-5', H-6')
3.32 and 3.64 (each s, 3 H and 6 H, OMe-4, OMe-6,
OMe-4'), 5.70 (d, J = 2.5 Hz, 1 H, H-5), 5.88 (d,
 J = 2.5 Hz, 1 H, H-3), 6.61 (d, J = 9 Hz, 2 H, H-3', H-5'),
7.33 (d, J = 9 Hz, 2 H, H-2', H-6'), 11.72 (s, 1 H, OH, D₂O exchangeable)
3.50, 3.80, 3.86 and 3.90 (each s, each 3 H, OMe-4,
OMe-6, OMe-3', OMe-4'), 5.93 (d, J = 2.5 Hz, 1 H, H-5),
6.14 (d, J = 2.5 Hz, 1 H, H-3), 6.82 (d, J = 9 Hz,
1 H, H-5'), 7.21 (dd, J = 9 Hz and 2.5 Hz, 1 H, H-6'),
7.57 (d, J = 2.5 Hz, 1 H, H-2'), 11.96 (s, 1 H, OH, D₂O exchangeable)

Experimental

Preparation of 2-Hydroxy-4,6,4'-trimethoxybenzophenone (7), General Procedure

To a cooled and stirred solution of 1,3,5-trimethoxybenzene (5 g, 0.029 mol) in absolute diethyl ether (15 ml) and anhydrous aluminium chloride (9.94 g, 0.074 mol) was added 4-methoxybenzoyl chloride (5.05 g, 0.029 mol) over 1 h. The stirring continued for further 4 h and the reaction mixture was left overnight. Working up of the reaction gave 7 (6.01 g) which crystallised from benzene—petroleum ether as colourless needles, m.p. 150–151°.

*Preparation of 7-Methoxy-4-phenyl-2*H*-1-benzopyran-2-one (2), General Procedure*

A mixture of 2-hydroxy-4-methoxybenzophenone (2 g, 0.0088 mol) and ethoxycarbonylmethylenetriphenylphosphorane (4.57 g, 0.0132 mol) was refluxed in anhydrous benzene (25 ml) according to the time mentioned in Table 1. After evaporation of the solvent, the residue was chromatographed over silica gel. Elution of the column with benzene : petroleum ether (3 : 1) gave 2 (1.54 g). It was crystallised from benzene—petroleum ether as colourless prisms, m.p. 146–147°.

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References

- ¹ Donnelly D. M. X., Neoflavanoids. In: The Flavonoids (Mabry T. J., Harborne J. B., Mabry H., eds.), p. 801. London: Chapman & Hall. 1975.
- ² Monache G. D., Botta B., Neto A. S., Lima R. A. D., Phytochemistry **22**, 1657 (1983).
- ³ Ahluwalia V. K., Seshadri T. R., J. Chem. Soc. **1957**, 970.
- ⁴ Donnelly D. M. X., Thompson J. C., Whalley W. B., Ahmed S., J. Chem. Soc. Perkin I **1973**, 1737.
- ⁵ Ollis W. D., Gotlieb O. R., Chem. Commun. **1968**, 1396.
- ⁶ Mukerjee S. K., Saroja T., Seshadri T. R., Indian J. Chem. **7**, 671 (1969).
- ⁷ Mali R. S., Yadav V. J., Synthesis **1977**, 464.
- ⁸ Ahluwalia V. K., Singh R. P., Tripathi R. P., Monatsh. Chem. **115**, 765 (1984).
- ⁹ Kauffmann H., Pannwitz P., Ber. **43**, 1208 (1910).
- ¹⁰ Ahluwalia V. K., Mehta A. C., Seshadri T. R., Proc. Ind. Acad. Sci. **45A**, 15 (1957).
- ¹¹ Karrer P., Liechtenstein N., Helv. Chim. Acta **11**, 789 (1928).
- ¹² Ulibelen A., Kerr R. R., Mabry T. J., Phytochemistry **21**, 1145 (1982).